A certain natural product (represented as $\mathbf{S}$ in the schemes below) can be synthesized racemically as follows:
 cat. MeLi, PhOEt
210 C , Microwav e Irridiation

$\xrightarrow[\text { 2) Mel }]{\text { 1) } \mathrm{LiN}\left(\mathrm{SiMe}_{3}\right)_{2}} P \xrightarrow{\left(\mathrm{NH}_{4}\right)_{2} \mathrm{Ce}\left(\mathrm{NO}_{3}\right)_{8}} \mathrm{Q} \xrightarrow[\text { 2) } \mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}]{\text { 1) } \mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}} \mathrm{R} \xrightarrow{\mathrm{TsOH}}$ raC-S
$\mathrm{Cp}=-\mathrm{C} 5 \mathrm{H} 5, \mathrm{Bz}=-\mathrm{COPh}$,
Alternatively, the natural product can be enantioselectively synthesized as follows:


1. Draw the structures of intermediates $\mathbf{A}-\mathbf{R}$ (no stereochemistry required).

Hints:
The formula of $B$ is $\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$.
The formula of I is $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{3}$.
The NMR spectrum of $\mathbf{A}$ is as follows:
13C NMR: 155.02, 129.79, 121.09, 115.48
1H NMR: $7.240(1 \mathrm{H}), 6.931(2 \mathrm{H}), 6.838(2 \mathrm{H}), 5.35\left(1 \mathrm{H}\right.$, exchanges with $\left.\mathrm{D}_{2} \mathrm{O}\right)$
Compound $\mathbf{R}$ contains 2 aromatic cycles.
rac-S is a more stable isomer of Compound $\mathbf{R}$.

| A | B | C | D |
| :--- | :--- | :--- | :--- |
|  |  | F |  |

2. Draw the structures of Z-W (with stereochemistry) and $\mathbf{S}$ (you may draw rac-S without penalty).

Hints:
Compound $\mathbf{Z}$ has $\mathbf{R}$ configuration.
Compound $\mathbf{W}$ is tetracyclic and contains two oxygen atoms.

| $\mathbf{Z}$ |  | $\mathbf{Y}$ |
| :--- | :--- | :--- |
|  |  |  |

3. Draw the mechanism of the reaction between $\mathbf{Y}$ and $\mathbf{Z}$ that produces $\mathbf{X}$. What is this reaction known as?

