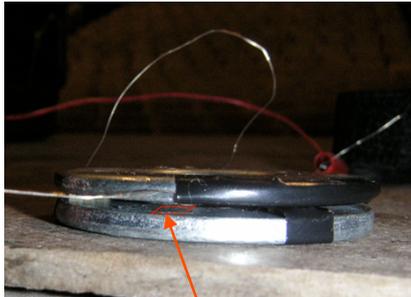
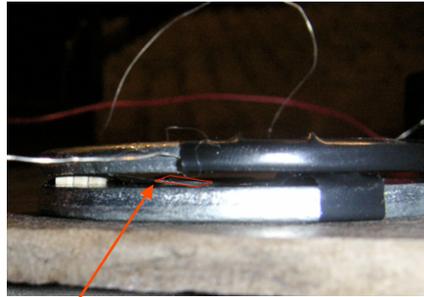


EF Experiment

Small V -
Foil lies on lower plate



Bigger V -
Foil jumps up and connects washers



Aluminum Foil

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EF Experiment

Bigger V -
Foil jumps up and connects washers

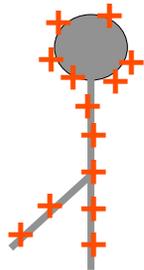


- Make sure that foil is clean, flat and not too smooth
- Start with foil flat on lower washer and min HVPS voltage
- Slowly turn up HVPS until foil jumps
- Foil jumps when $V = V_{\text{jump}}$ such that electrostatic force balances weight of foil
- Repeat experiment several times for each foil to get consistent V_{jump}
- Repeat with foils folded to have 1, 2, 3 layers

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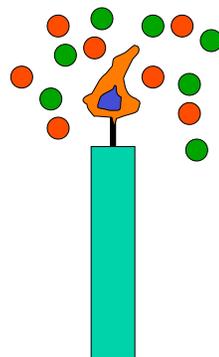
web.mit.edu/8.02x/www

Demo I



Electroscope

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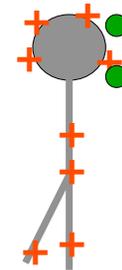


Charged Ions

web.mit.edu/8.02x/www

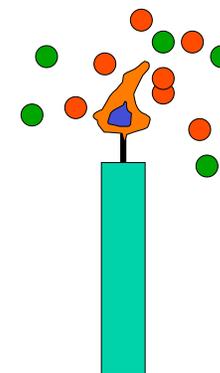
Demo I

Ions discharge
Electroscope



Electroscope

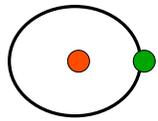
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Charged Ions

web.mit.edu/8.02x/www

Demo I



Neutral molecules:
Pos. and neg. charges move together -> No current!

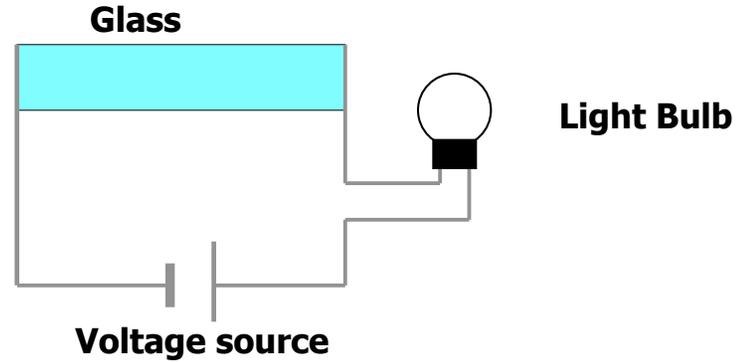


Ions:
Pos. and neg. charges move separately -> Current $|I| > 0$!

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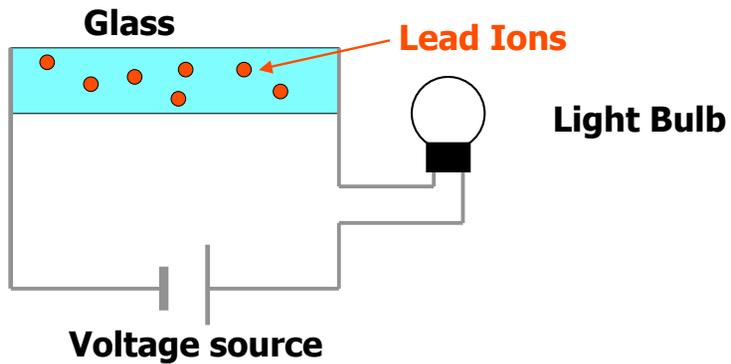
Demo II



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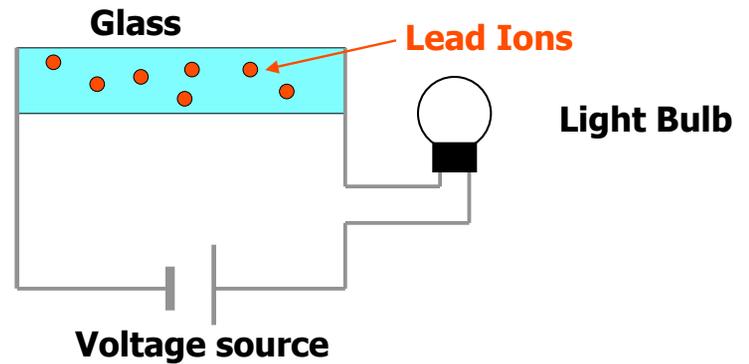
Demo II



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Demo II

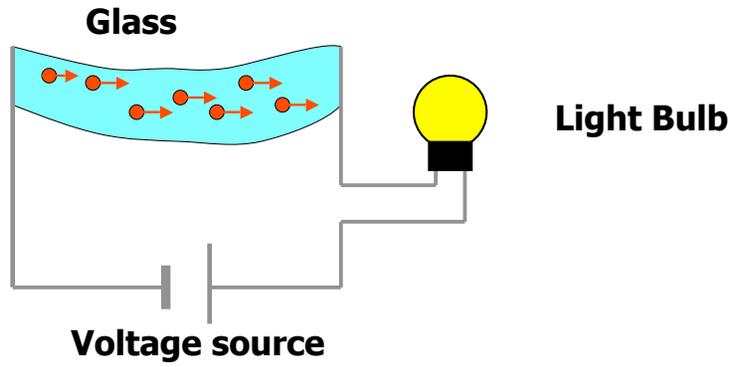


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Solid glass: Potential charge carriers are stuck!

Demo II



Molten glass: Charge carriers become mobile ->
Current flows -> Bulb lights up!